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## HOMEWORK - June 2023

- Problem I

$$\textcircled{1} \quad C = \underbrace{\alpha T}_{\text{Electrons}} + \underbrace{\beta T^3}_{\text{Phonons}} = C_{el} + C_{ph}$$

$C_T = \alpha + \beta T^3$  Then  $C_T = f(T^2)$  is a linear expression.

$$\textcircled{2} \quad \text{Potassium: } K : [\text{Ar}] 4s^1$$

$$C_{el} = \frac{\pi^2 N k_B}{2} \left( \frac{T}{\Theta_F} \right) = 1,668 \times 10^{-3} T$$

$$\text{So } \underline{\Theta_F^K = 24578 \text{ K}}$$

$$\text{Sodium: } Na : [\text{Ne}] 3s^1$$

$$C_{el} = \frac{\pi^2 N_A k_B}{2} \left( \frac{T}{\Theta_F} \right)_{Na} = 1,094 \times 10^{-3} T$$

$$\text{So } \underline{\Theta_F^{Na} = 37474 \text{ K}}$$

$\textcircled{3}$  Number of quantum states below energy  $E$ ,

$$\phi(E) = 2 \times \frac{\frac{4}{3} \pi \left( \frac{2mE}{\hbar^2} \right)^{\frac{3}{2}}}{\left( \frac{2\pi L}{L} \right)^3} = \frac{\sqrt{}}{3\pi^2} \times \left( \frac{2mE}{\hbar^2} \right)^{\frac{3}{2}}$$

Then density of states: 
$$g(E) = \frac{d\phi}{dE} = \frac{\sqrt{}}{2\pi^2} \left( \frac{2m}{\hbar^2} \right)^{\frac{3}{2}} E^{\frac{1}{2}}$$

④ The fermi energy ( $E_F$ ) corresponds to the energy of highest occupied state, then:

$$\phi(E_F) = N \quad \text{if } N \text{ is the total number of electrons.}$$

$$\phi(E_F) = \frac{V}{3\pi^2} \left( \frac{2m}{\hbar^2} \right)^{3/2} E_F^{3/2} = N$$

$$\Rightarrow E_F = \frac{\hbar^2}{2m} \left( 3\pi^2 \frac{N}{V} \right)^{2/3} = k_B \Theta_F$$

• Potassium:  $\frac{N}{V} = \frac{2}{a_K^3}$  with 2 atoms (electrons) per cubic cell in Bcc.

$$\Theta_F^K = \frac{\hbar^2}{2mk_B} \left( 3\pi^2 \times \frac{2}{a_K^3} \right)^{2/3} = \underline{23679 \text{ K}}$$

$$\text{Error} = \frac{\Delta \Theta_F^K}{\Theta_F^K} \times 100\% = \underline{\underline{3,6\%}}$$

This small discrepancy between experimental and theoretical value comes from the "Fermi gas" approximation made on valence electrons.

• Sodium:  $\Theta_F^{Na} = \underline{36552 \text{ K}}$

$$\frac{\Delta \Theta_F^{Na}}{\Theta_F^{Na}} \times 100\% = \underline{\underline{8,5\%}}$$

Same remark.

⑤ The Debye temperature of Solids can be deduced from the expression of phononic contribution to the heat capacity:

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$$C_{ph} = \frac{12\pi^4}{5} N k_B \left(\frac{T}{\Theta_D}\right)^3$$

Potassium:

$$C_{ph} = 2,75 \times 10^{-3} T^3$$

then  $\Theta_D^K = \underline{\underline{89 \text{ K}}}$

Sodium:  $\Theta_D^{Na} = \underline{\underline{153 \text{ K}}}$

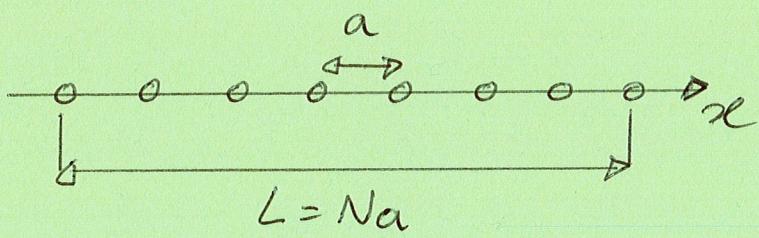
⑥ The highest is the Debye temperature, (= highest is the Debye frequency since  $\omega_D = k_B \Theta_D / \hbar$ ), the highest is the strength of the interatomic bonds. So the "hardest" is the Solid.

Then a so-called "soft" solid is characterized by a low Debye freq. and temperature.

As an example, the C-diamond considered as part of the hardest materials is charact. by a Debye temp. of about 1900 K.

As a result, the heat capacity rises very quickly with increasing temperature for the case of "soft" materials (low  $\Theta_D$ ) and rises very slowly for "harder" materials (high  $\Theta_D$ ). [4]

### Problem II :



① free electrons along this line of atoms are described by wave-functions of type:  $\psi_{\text{free}} = Ae^{ikx}$

Then, by applying cyclic boundary conditions:

$$\psi(0) = \psi(L) = Ae^{ikL} = A$$

$$\Rightarrow kL = n\pi$$

$$\Rightarrow \boxed{k = \frac{n\pi}{L}}$$

Quantization of wave-vector.

Application of Schrödinger equation gives the energy of any electron described by a given quantum state  $(\psi_k)$ .

$$\boxed{E_k = \frac{\hbar^2 k^2}{2m}}$$

Dispersion relation

Since  $L$  electrons can be accommodated on one quantum state, the highest occupied state corresponds to  $k = \frac{N}{L} \times \frac{2\pi}{L} = \frac{N\pi}{L}$

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$$\text{Then: } E_F = \frac{\hbar^2}{2m} \times \left(\frac{N\pi}{L}\right)^2 = \frac{\hbar^2}{2m} \left(\frac{\pi}{a}\right)^2 = 4,18 \text{ eV}$$

$$\Theta_F = \frac{E_F}{k_B} = \underline{48550 \text{ K}}$$

example: Li ( $Z=3$ )

$$E_F = 4,18 \text{ eV} \quad a_{Li} = 3,25 \text{ \AA}$$

$$\Theta_F = 55100 \text{ K}$$

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Number of quantum states below a given value of  $k$ :

$$\phi(k) = \frac{k}{\frac{2\pi}{L}} = \frac{Lk}{2\pi} = \frac{Na}{2\pi}$$

$$\text{Then } g(k) = \frac{d\phi(k)}{dk} = \frac{aN}{2\pi}$$

$$g(E) dE = g(k) dk$$

$$\Rightarrow g(E) = g(k) \frac{dk}{dE} = \frac{aN}{2\pi} \times \frac{1}{\frac{dE}{dE}} \left( \sqrt{\frac{2me}{\hbar^2}} \right)$$

$$\Rightarrow g(E) = \frac{aN}{2\pi} \times \frac{1}{\hbar} \sqrt{\frac{2m}{E}} \times \frac{1}{\sqrt{2E}}$$

$$\Rightarrow g(E) = \frac{Na}{2\pi} \times \frac{1}{\hbar} \sqrt{\frac{m}{2E}} \times 2 \quad \text{if spin and Pauli principle is used.}$$

$$g(E) = AE^{-\frac{1}{2}}$$

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$$U_0 = \int_0^{E_F} E g(E) dE = A \int_0^{E_F} E^{1/2} dE$$

$$\Rightarrow U_0 = \frac{2}{3} A E_F^{3/2}$$

with

$$N = \int_0^{E_F} A E^{-1/2} dE = 2 A E_F^{1/2}$$

$$\Rightarrow A = \frac{N}{2} E_F^{-1/2}$$

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$$U_T = \int_0^{+\infty} E g(E) f(E) dE *$$

$$\text{So } U_0 = \frac{N}{3} E_F$$

Using Sommerfeld approximation:

$$U_T = \int_0^{\mu} E g(E) dE + \frac{\pi^2}{6\beta^2} \left( \frac{A}{2} E_F^{-1/2} \right)$$

$$\text{with } \phi(E) = E g(E) = A E^{1/2}$$

$$\phi'(E) = \frac{A}{2} E^{-1/2}$$

$$\text{Then: } U_T = U_0 + \int_{E_F}^{\mu} A E^{1/2} dE + \frac{\pi^2}{6\beta^2} \left( \frac{A}{2} E_F^{-1/2} \right)$$

$$U_T \approx U_0 + A E_F^{1/2} (\mu - E_F) + \frac{A \pi^2}{18\beta^2} E_F^{-1/2}$$

$$\text{At low T: } \mu - E_F \approx 0 \quad \text{and } U_T \propto T^2$$

and heat capacity  $C$  proportional to  $T$

$$C = \frac{dU}{dT} \propto T$$

### Problem III :

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3D semi-metal DOS :

$$\begin{cases} g(E) = A \sqrt{E_F - E} & \text{for } E < E_F \\ g(E) = A \sqrt{E - E_F} & \text{for any } E > E_F \\ g(E) = 0 & \text{at } E = E_F \end{cases}$$

$$\textcircled{1} \quad N = \int_0^{E_F} g(E) dE = A \int_0^{E_F} \sqrt{E_F - E} dE$$

$$\Rightarrow N = A \left[ -\frac{2}{3} (E_F - E)^{\frac{3}{2}} \right]_0^{E_F} = \frac{2}{3} A E_F^{\frac{3}{2}}$$

Then  $\boxed{A = \frac{3N}{2} E_F^{-\frac{3}{2}}}$

$$\textcircled{2} \quad U_0 = \int_0^{E_F} E g(E) dE = A \int_0^{E_F} E \sqrt{E_F - E} dE$$

change in variables:

$$\begin{cases} X = E_F - E \\ E = E_F - X \\ dE = -dX \end{cases}$$

$$\begin{aligned} U_0 &= -A \int_{E_F}^0 (E_F - X) \sqrt{X} dX = A \int_0^{E_F} (E_F - X) \sqrt{X} dX \\ &= A \int_0^{E_F} E_F \sqrt{X} dX - A \int_0^{E_F} X^{\frac{3}{2}} dX \end{aligned}$$

$$U_0 = \frac{2}{3}AE_F \left[ x^{\frac{3}{2}} \right]_0^{E_F} - \frac{2A}{5} \left[ x^{\frac{5}{2}} \right]_0^{E_F}$$

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$$\Leftrightarrow U_0 = \frac{2}{3}AE_F^{\frac{5}{2}} - \frac{2}{5}AE_F^{\frac{5}{2}} = \frac{4}{15}AE_F^{\frac{5}{2}}$$

③  $N_{CB}^{el} = \int_{E_F}^{+\infty} g_{CB}(E) f_{CB}(E) dE$

with  $f_{CB}(E) = \frac{1}{e^{\beta(E-\mu)} + 1} \ll 1$

↳ Boltzmann distribution

$$f_{CB}(E) \approx e^{-\beta(E-\mu)}$$

Then:

$$N_{CB}^{el} = \int_{E_F}^{+\infty} A \sqrt{E-E_F} e^{-\beta(E-\mu)} dE$$

Again chg. in variables

$$X = \beta(E-E_F)$$

etc...

$$N_{CB}^{el} = \frac{A\sqrt{\pi}}{2} (k_B T)^{\frac{3}{2}} e^{-\beta E_F} e^{\beta \mu}$$

④  $N_{VB}^h = A \int_{-\infty}^{E_F} \sqrt{E_F-E} e^{\beta(E-\mu)} dE$

$$N_{VB}^h = A \int_{+\infty}^0 -\sqrt{\frac{x}{\beta}} e^{-x} e^{\beta E_F} e^{-\beta \mu} \frac{dx}{\beta}$$

$$\begin{cases} X = \beta(E_F - E) \\ dX = -\beta dE \\ E = -\frac{X}{\beta} + E_F \\ dE = -dx/\beta \end{cases}$$

$$N_{VB}^h = \int_0^{+\infty} \sqrt{x} e^{-x} dx \left( \frac{A}{\beta^{\frac{3}{2}}} e^{\beta E_F} e^{-\beta \mu} \right)$$

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$$= \sqrt{\frac{\pi}{2}}$$

finally :

$$N_{VB}^h = \frac{A\sqrt{\pi}}{2} (k_B T)^{\frac{3}{2}} e^{\beta E_F} e^{-\beta \mu}$$

⑤ When one electron is promoted to a quantum state in the conduction band, a vacancy or a hole is created in the valence band so that:

$$N_{VB}^h = N_{CB}^e$$

⑥ At low  $T$  ( $k_B T \ll E_F$ ) and  $\mu \approx E_F$

$$\text{So } N_{CB}^e = \frac{A\sqrt{\pi}}{2} (k_B T)^{\frac{3}{2}} e^{-\beta E_F} e^{\beta \mu} \approx \frac{A\sqrt{\pi}}{2} k_B^{\frac{3}{2}} T^{\frac{3}{2}}$$

$$N_{CB}^e \propto T^{\frac{3}{2}}$$

⑦ If  $N_{VB}^h = N_{CB}^e$  then  $e^{\beta E_F - \beta \mu} = e^{-\beta E_F} e^{\beta \mu}$

$$\Leftrightarrow \beta(E_F - \mu) = \beta(\mu - E_F)$$

$$\Leftrightarrow \mu = E_F$$

So again  $N_{CB}^e \propto T^{\frac{3}{2}}$  no activation energy.