

Part I: Just a little theory

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① Effective mass of the charge carriers proportional to the effective potential  $\Rightarrow$  the highest the interaction with nuclei (positive ions), the highest is the mass, so the lowest will be the mobility.

② Dispersion relation:  $E_k = \frac{\hbar^2}{2m_{\text{eff}}} k^2$   
with  $m_{\text{eff}} = \frac{m^2 a^2 |V|}{\hbar^2}$

③  $\frac{d^2 E_k}{dk^2} = \frac{\hbar^2}{m_{\text{eff}}} \Leftrightarrow m_{\text{eff}} = \frac{\hbar^2}{d^2 E_k / dk^2}$

④ The highest is the local curvature ( $\frac{d^2 E}{dk^2}$ ) the lowest the effective mass. In point C, the curvature is lower and effective mass higher.

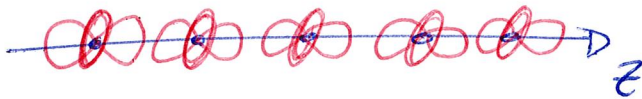
⑤ Lowest mass  $\Rightarrow$  higher mobility -  
(point A) (point A)

⑥ Band gap between A and B:  $\Delta E = 0,7 \text{ eV}$

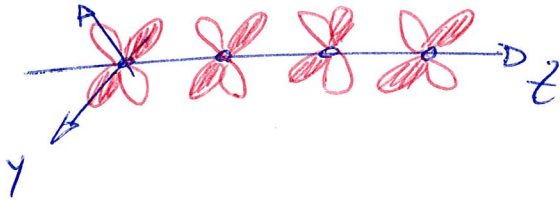
(7) The higher the curvature, the ~~the~~ lowest the DOS. DOS is high at point c. 2

Part II: A chemist view ---

①  $d_{z^2}$  large overlap  $\Rightarrow$  broad band  
( $\sigma$ ) (high dispersion)



$d_{x^2-y^2}, d_{xy}$  small overlap ( $\delta$ )  $\Rightarrow$  narrow band  
(small dispersion)



②  $d_{z^2}$  Bonding state for  $k=0$   $k = \frac{\pi}{a}$   
Anti-bonding

$p_z$  Anti-bonding state for  $k=0$    
Bonding state for  $k = \frac{\pi}{a}$

③  $N$  states with  $N$  the number of atoms (groups)

$$k = n \frac{2\pi}{L} = n \frac{2\pi}{Na}$$

- full length of 1<sup>st</sup> Brillouin zone =  $\frac{2\pi}{a}$

- number of states  $\frac{2\pi/a}{2\pi/Na} = N$  or  $2N$  spins

Part III. <sup>④</sup> The lowest the distance, the broadest the band. 3

① Cours - Bands and conductivity.

② The highest is the dispersion, the highest is the overlap between orbitals = the highest the coupling and mobility of electrons = the highest the conductivity. CHEMIST

Current density  $\vec{j} = -e \int \frac{d^3k}{4\pi^3} \cdot \frac{dE(\vec{k})}{d\vec{k}}$   
The highest the dispersion, the highest the integral (sum) over  $d^3k$  of  $\frac{dE(\vec{k})}{d\vec{k}}$  = the highest the conductivity. Physicist

③ METAL.

